

TITRATION

Meaning: Titration is a laboratory technique used to determine the concentration of a substance in a solution by adding a reactant of known concentration until a reaction is complete.

1. Titration is a critical technique in chemistry used to determine the concentration of a solution.
2. During the experiment, the student carefully performed a titration to find the exact amount of acid in the sample.
3. The endpoint of the titration was indicated by a color change of the pH indicator in the solution.
4. Accurate titration requires precise measurements and careful observation to achieve reliable results.
5. In the laboratory, titration is often used to analyze the purity of a substance or the strength of an acid or base.

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